

## Naming Acids

### ● Binary Acids

- prefix HYDRO and suffix IC ACID
- Example: HCl = *hydrochloric acid*

### ● Oxyacids

oxidation state	polyatomic ion			acid name	
	example	prefix	suffix	prefix	suffix
two less than most common	$\text{ClO}^{-1}$	hypo	ite	hypo	ous
one less than most common	$\text{ClO}_2^{-1}$	-	ite	-	ous
most common	$\text{ClO}_3^{-1}$	-	ate	-	ic
one more than most common	$\text{ClO}_4^{-1}$	hyper	ate	per	ic

- Example:  $\text{HNO}_3$  = hydrogen nitrate;  $\text{HNO}_3(aq)$  = nitric acid