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## More Names and Formulas!

Write the correct name of the compound on the space provided for each of the formulas listed below.

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|--|-----------|
| 1. $\text{SnCl}_4$                       | 1. _____  |
| 2. $\text{Ca}(\text{ClO}_3)_2$           | 2. _____  |
| 3. $\text{NH}_4\text{NO}_2$              | 3. _____  |
| 4. $\text{Pb}(\text{SO}_4)_2$            | 4. _____  |
| 5. $\text{N}_2\text{O}_5$                | 5. _____  |
| 6. $\text{MnF}_7$                        | 6. _____  |
| 7. $\text{Mg}(\text{HCO}_3)_2$           | 7. _____  |
| 8. $\text{AgBr}$                         | 8. _____  |
| 9. $\text{NaI}$                          | 9. _____  |
| 10. $\text{Ni}(\text{CH}_3\text{COO})_2$ | 10. _____ |

Write the correct formula on the space provided for each of the compounds listed below.

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|------------------------|-----------|
| 11. Iron III sulfite   | 11. _____ |
| 12. Potassium chlorate | 12. _____ |
| 13. Zinc acetate       | 13. _____ |
| 14. Aluminum hydride   | 14. _____ |
| 15. Nickel II sulfide  | 15. _____ |
| 16. Carbon IV oxide    | 16. _____ |
| 17. Copper I sulfite   | 17. _____ |
| 18. Calcium bromide    | 18. _____ |
| 19. Silver sulfide     | 19. _____ |
| 20. Tin IV phosphate   | 20. _____ |