

Periodic Table of the Elements

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1	<table border="1"> <tr> <td>1.00794</td> <td>+1</td> <td colspan="8"></td> <td>-1</td> </tr> <tr> <td>H</td> <td></td> <td colspan="8"></td> <td></td> </tr> <tr> <td>1</td> <td></td> <td colspan="8"></td> <td></td> </tr> </table>										1.00794	+1									-1	H											1											<table border="1"> <tr> <td>4.00260</td> <td>0</td> </tr> <tr> <td>He</td> <td></td> </tr> <tr> <td>2</td> <td></td> </tr> </table>						4.00260	0	He		2		18																																																																																																						
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*denotes the presence of (2-8-) for elements 72 and above

**The systematic names and symbols for elements of atomic numbers 113 and above will be used until the approval of trivial names by IUPAC.

Source: *CRC Handbook of Chemistry and Physics*, 91st ed., 2010–2011, CRC Press